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~~Multiple choice Question 1.4 – 1.10~~

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Kinetic Energy, Gravitational

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Please turn over QUESTION 5

(Start on a new page.) Zinc granules are added to 100 cm³ of a 0,2 mol · dm⁻³ hydrochloric acid solution in an Erlenmeyer flask.

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The equation for the reaction that takes place is: $\text{Zn(s)} + 2\text{HCl(aq)} \rightarrow \text{ZnCl}_2(\text{aq}) + \text{H}_2(\text{g})$

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2014 WATER BATH 1. Pour about 100 cm³ of water into the 250 cm³ beaker. 2. Carefully lower the tube into the beaker so that it stands upright. 3. Heat the beaker gently on a tripod and gauze until the water begins to boil, and then stop the heating. 4.

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Learning Outcomes and
Assessment Standards

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12.1.1: Design, plan and conduct a
scientific inquiry to collect data

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2014 systematically with regard to accuracy, reliability and the need to control variables.

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4.5 A variable resistor, an

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ammeter, a battery of emf 12 V and voltmeters V 1 and V 2 are connected as shown in the diagram below. When the switch is open, the readings on voltmeters V1 and V 2 respectively are ... Reading on V 1 Reading on V 2 A 12 V 0 V B 12 V 12 V C 0 V 0 V D 0 V 12 V

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2014
(3) [15] TOTAL SECTION A: 35
A R V1 V2

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3.4 From the graph above, deduce and then write down the NAME of the hydride: 3.4.1 With the weakest intermolecular forces (1)

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~~2014~~ 3.4.2 With hydrogen bonds
between molecules (1)

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