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The science of chemistry rests on two well-established principles: the conservation of matter and the conservation of energy. Chemists think in the highly visual nanoscopic world of atoms and molecules, but they work in the tangible world of macroscopic laboratory apparatus. These two approaches to the chemical sciences cannot be divorced.

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The Student Solutions manual, authored by Wade Freeman of the University of Illinois at Chicago, contains solutions to the odd numbered problems.

Appropriate for 2-semester or 3-quarter general chemistry courses. General Chemistry: Principles and Modern Applications is recognized for its superior problems, lucid writing, and precision of argument. This edition introduces a number of innovative features—including new Feature Problems, new follow-up Practice Exercises to accompany every in-chapter Example, and a number of new Focus On application boxes.

The 7th Edition of Gary Christian's Analytical Chemistry focuses on more in-depth coverage and information about Quantitative Analysis (aka Analytical Chemistry) and related fields. The content builds upon previous editions with more enhanced content that deals with principles and techniques of quantitative analysis with more examples of analytical techniques drawn from areas such as clinical chemistry, life sciences, air and water pollution, and industrial analyses.

Written for calculus-inclusive general chemistry courses, Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications. Unlike other texts, it begins with a detailed picture of the atom then builds toward chemistry's frontier, continually demonstrating how to solve problems, think about nature and matter, and visualize chemical concepts as working chemists do. Flexibility in level is crucial, and is largely established through clearly labeling (separating in boxes) the calculus coverage in the text: Instructors have the option of whether to incorporate calculus in the coverage of topics. The multimedia integration of Chemical Principles is more deeply established than any other text for this course. Through the unique eBook, the comprehensive Chemistry Portal, Living Graph icons that connect the text to the Web, and a complete set of animations, students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding.

In its Seventh Edition, this acclaimed Clinical Chemistry continues to be the most student-friendly clinical chemistry text available. This edition not only covers the how of clinical testing but also places greater emphasis on the what, why, and when in order to help today's students fully understand the implications of the information covered, as well as the applicability of this crucial topic in practice. With clear explanations that strike just the right balance of analytic principles, techniques, and correlation of results with disease states, this edition has been fully updated with the latest information to help keep today's students at the forefront of today's science. New case studies, practice questions, and exercises provide ample opportunities to review and apply the topics covered through the text.

Acclaimed by students and instructors alike, Foye's Principles of Medicinal Chemistry is now in its Seventh Edition, featuring updated chapters plus new material that meets the needs of today's medicinal chemistry courses. This latest edition offers an unparalleled presentation of drug discovery and pharmacodynamic agents, integrating principles of medicinal chemistry with pharmacology, pharmacokinetics, and clinical pharmacy. All the chapters have been written by an international team of respected researchers and academicians. Careful editing ensures thoroughness, a consistent style and format, and easy navigation throughout the text.

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